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Answer will vary; because the normality of a solution equals the equivalent weight of solute per liter of solution, a solution containing 1 equivalent mass of an acid will be a 1 N solution. In your own words, explain why a 1 M HCl solution and a 1N HCl solution have the same concentration, but a 1 M HCl solution and a one N H<sub>2</sub>SO<sub>4</sub> solution do not.

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